

# New Biodiesel Process and Technology Provided by Us

**Hirai Corp.,Inc.**

**Akira Hirai**

**Nov.14th,2011**

# What is biodiesel and its benefit in general ?

**Biodiesel(BD) fuel** has been tested extensively for emissions and compatibility with existing engines and meet fuel property standards.

Biodiesel met include:

- Development and meet of a fuel standard (ASTM,EN,JIS)
- Engine manufacturer's acceptance
- Positive public image and acceptance(Renewable/Ecological fuel)
- Contains no petroleum, but can be blended at any level with petroleum-based diesel to create a biodiesel blended fuel.
- Burns in compression-ignition (diesel) engines without having to perform major modifications to the engine.
- Simple to use, biodegradable, non-toxic and free of sulfur and aromatics,not like petroleum

# Biodiesel Fuel Advantages

- A more environmentally friendly alternative to petroleum diesel fuel
- An alternative to reduce greenhouse gas emissions such as CO<sub>2</sub>, carbon monoxide, particulate matter and hydrocarbon emissions. ,but increase NO<sub>x</sub> emission slightly.
- Can be made simply from used oil(WVO/UCO) or unused vegetable oils(VO),off-specification waste oils derived from the food processing industry or from fats, tallow and animal fats and non-food vegetable or tree seed oil(Jatropha,Castor Oil,etc.).
- Used as diesel engine fuel(cars/heavy duty construction machines) w/o modification and also for boiler/generator burning fuels(Bioheat).
- Compatible with the existing fuel distribution & infrastructure facilities.
- Biodiesel reduces emissions by more than 80-90%(with natural gas based methanol,but almost 100% with renewable ethanol based biodiesel).



**Biodiesel said to be poor winter cold weather performance in general.. However, can be used up to -40 or less deg.C(See above),if treated properly by **our technology**.**

# Biodiesel Basics and its Problems

- Reaction(called Transesterification) takes place chemically such as  
 $1 \text{ (VO)} + 3 \text{ (Methanol)} \Rightarrow 3 \text{ (BD)} + 1 \text{ (Glycerin)}$   
BD: Biodiesel
- Commonly,traditional (homogeneous) lye (NaOH,KOH) used as catalyst and made BD as product ,and glycerin,soap,etc as byproduct.
- Need purification to remove byproducts through water-washing,or newer dry-processing.
- As the state of arts of technology for BD, heterogenous(solid) catalysts(Metal Oxides, Enzymes) processes are just recently introducing and higher purity BD as well as clean chemical grade-glycerin are produced(no-catalyst and no-soap byproduct separation).
- Metal oxide processes mostly need higher temperature( pressure) to speed-up reaction kinetics



As extensive evaluation works were performed in past a few years, **solid catalysts** were future way as our **next-generation biodiesel process**.

Results become **EnZymatic catalyst** were selected as the best among them.

# Main differences between biodiesel made using chemical Lye and EnZyme Solid Catalyst (1)

The differences are highlighted in the table below.

## EnZymatic Process

- **Reaction temperature: low as 10-35°C.**
- **Feedstock(FFA:Free Fatty Acid): Any purity of FFA feedstocks up to 100% .**
- **No effect on biocatalysis at water content <5%,so that low grade methanol/ethanol can be used.**
- **No-Soap byproduct.**
- **Biodiesel yield >98+% .**
- **Glycerin quality is transparent salt free & high quality.**
- **No catalyst removal and may no or only simplified Dry washing depending on fuel purity requirements.**

## Conventional Lye Process

- **55-70°C (higher energy).**
- **Reduced up to 2~3%(by acid or lye treatment for greater than 3%FFA) .**
- **Need pre-treatment de-hydration /FFA removal/higher water-free methanol.**
- **Makes soap with FFA.**
- **Typically, 96%(98%+ by our MSR:Milli-seconds reactor).**
- **Blackish-brown; pH >7; low quality; contains salt and others.**
- **Final product requires Wet/Dry washing to remove catalyst/soap.**

## Main differences between biodiesel made using chemical Lye and EnZyme catalyst (2)

### Enzymatic Process

- **Recyclable Catalyst for one year**
- **Methanol recovery :No or less excess methanol,which is enough for water content <5%.**
- **Waste generation:Extremely low**
- **Catalyst requirement /Ton:0.3 kg**
- **Cost of catalyst/kg of BD:\$0.05-0.07**
- **Capital cost:Low. Roughly 50% for conventional technologies**
- **Environment: Non-toxic**
- **Other operating costs:Very low**

### Conventional Process

- One time for each reaction
- Large excess of methanol (water content prohibited) . Requires stripping from final product
- Significant waste generation: catalyst, glycerin, soap,water
- 13-15 kg
- \$0.09-0.11
- High in general(Very low by our MSR;milli-Seconds Reactor)
- Toxic chemicals
- High(Low by MSR)

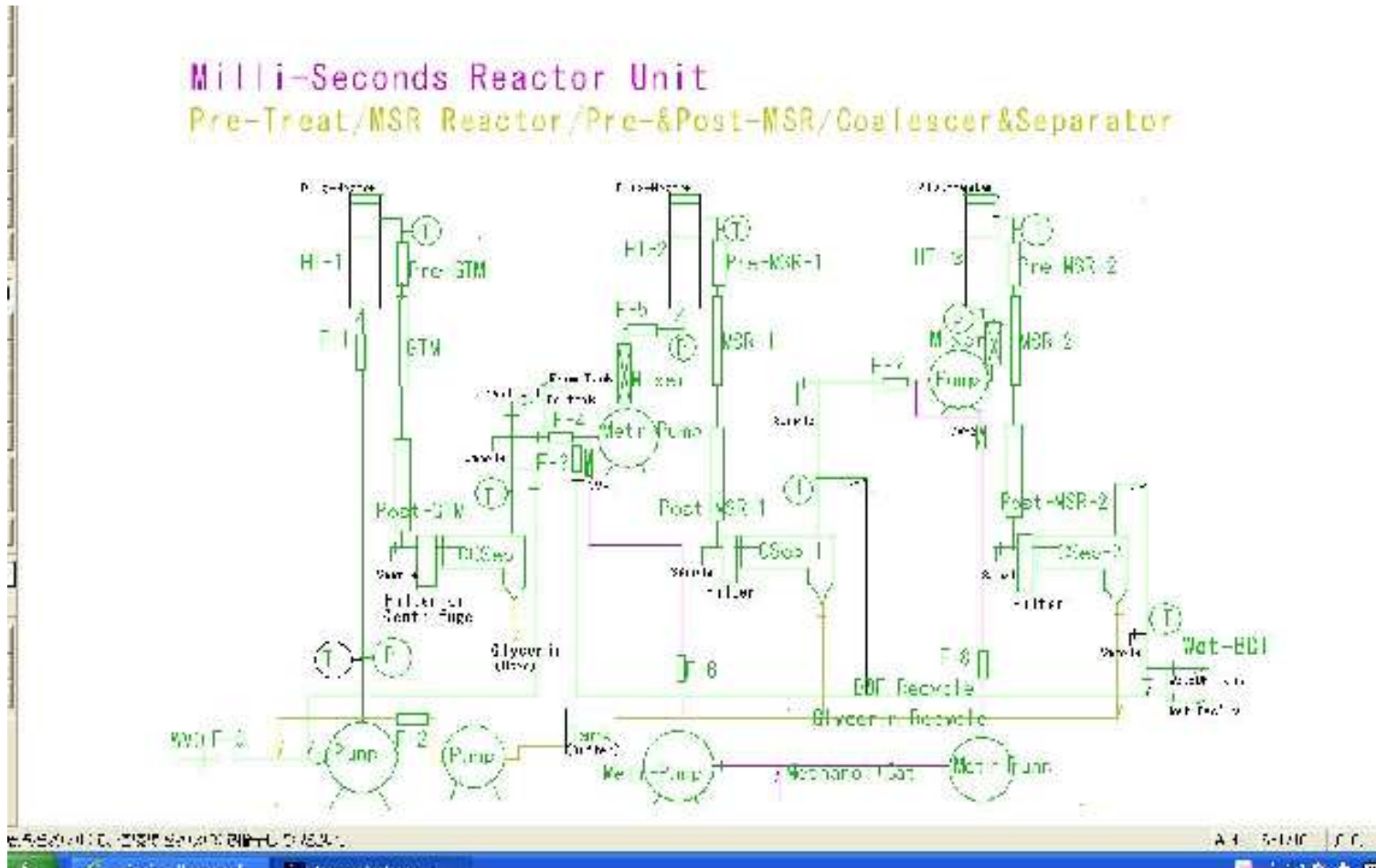
# The State of Arts, Newest Lye process, still revolutional process

- Typical lye based process need 2~6 hours of reaction time in batch mode for small to medium sized processor(100~3,000L).
- Also typically, need 1-3 hours of residence time for continuous process(large reactor size and cost-up), mostly applied for large plant above 5,000 or 10,000 L/day as minimum .
- Our newest state of arts, lye process (MSR: Milli-Seconds Reactor) has capable to :
  - +complete reaction within 300 to 600 milli-seconds(0.3-0.6seconds).
  - +Lowest cost(plant construction and operating expence) and shown small sized process(right) has 5,000-10,000L per day of capacity(dual 3/8" dia x 17" length).
  - +Typical methanol requirements are only 15-16%, which is lowest (Another process needs 20~22% typically).





# Lye Based Biodiesel MilliSeconds Reactor(MSR) Process Flow





# Glycerin byproduct usage of lye catalyst based process

## 1)As chemical feed stocks

- Base lye trans-esterified reaction (by additional distillation, cost-up).
- Higher purity glycerin by newest & state of art solid catalyst process without such complex operations.

## 2)As disposal or low-grade

- Boiler fuels mixed with petroleum fuel oils.
- Animal feed stocks (Up to 20%). See picture.

## 3)As medium value stocks

- Soap making feed as shown.
- Candle feed ,and others.



# The State of Arts, Newest EnZymatic Biodiesel Process

Solid catalyst based processes (metal oxide, EnZyme as Catalysts) are still very little in commercialization and ours is a top runner.

## Back-ground

- Capable multi-feed stocks( due to feed shortage and have to explore low-grade FFA oil), and non-food feed stocks(jatropha, palm residue oils, which contain higher fatty acid). Don't applicable by current lye base processes beyond FFA 3-4% .

- Need cost reduction pressure as whole

For plant construction(simple process and lowest cost).

For lowest energy requirement(30~35 deg.C).

For lowest methanol consumption (~15%), which occupy 60~65% of BD operating cost.

and also used or lower grade alcohol(~80%) can be used.

Higher grade glycerin (only by demethanol and dehydration ), which contributes to reducing BD cost more (Revenue by sale).



# Proposed EnZymatic Technology

## Overview

- **Breakthrough development of modified and methanol-resistant enzymes enabling industrial use of immobilized enzymes in the process of biodiesel production..**
- **Adaptation of enzymes (lipases) for use in organic synthesis.**
- **Immobilization of enzymes for industrial applications, which is achieved by binding the enzymes to specific solid carriers(resin).**
- **Technology overcomes the major drawbacks of enzyme catalysts in biodiesel production such as:**
  - +High cost of the enzymes and slow reaction rate.**
  - +Use of solvent to enhance the contact between oil and methanol for speed-up reaction rate(Highly active in solvent-free systems)..**
  - +The methanol-resistant modified EnZymes developed(Mostly,has to be used higher cost ethanol).**
  - +Tolerate low-grade raw materials(oil,methanol or also ethanol,if available) and cost less comparing others.**
  - +Can be used in both multiple batch&continuous processes for long time cycles.**



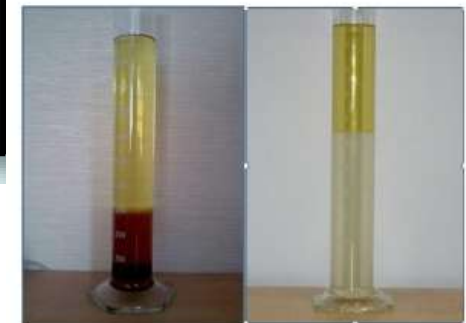
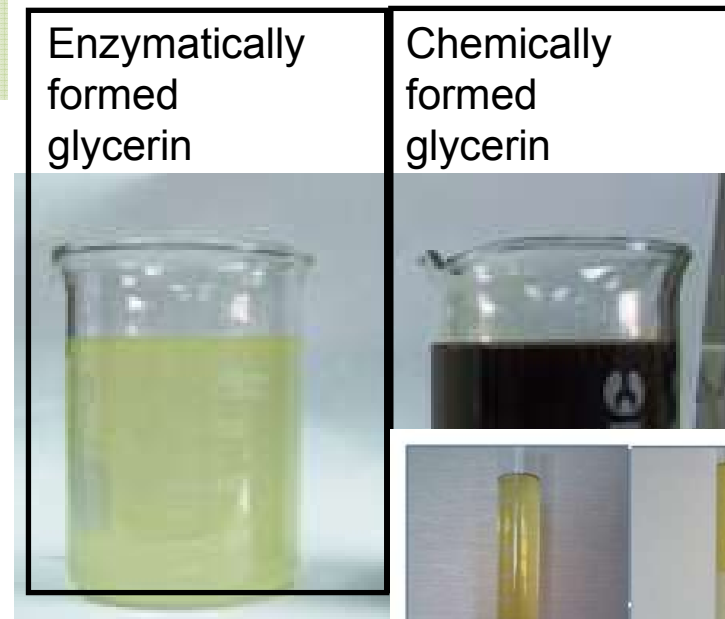
# Enzymatic Biodiesel Analysis Data from Soybean Oil

	Units	Test Method	Europe	USA	Results
Specification			EN 14214	ASTM D6751	EnZymeCatalyst
Density 15° C	g/cm <sup>3</sup>	ASTM D 1298	0.86-0.90		0.885
Viscosity 40° C	mm <sup>2</sup> /s	ASTM D445	3.5-5.0	1.9-6.0	4.2
Distillation	% @ ° C			90%,360° C	
Flash Point	° C	ASTM D93	120 min	130 min	180
Sulphur	mg/kg	ASTM D2622	10 max	15 max	<10
CCR 100%	% mass			0.05 max	400
Carbon Residue	% mass	ASTM D4530	0.03 max		<0.3
Sulphated Ash	% mass	ASTM D874	0.02 max	0.02 max	
Water	mg/kg	ASTM D2709	500 max	500 max	320
Total Contamination	mg/kg	ASTM D5452	24 max		<10
Max CU Corrosion	3h/50° C	ASTM D130	1	3	<1
Oxidation Stability [110° C]	Hours	EN 14112	6 min		
Cetane Number		ASTM D613	51 min	47 min	
Acid Value	mg KOH/g	ASTM D664	0.5 max	0.8 max	0.22
Methanol	% mass	EN 14110	0.2 max		<0.01
Ester Content	% mass	EN 14103	96.5 min		96.6
Monoglyceride	% mass	EN 14105	0.8 max		0.538
Diglyceride	% mass	EN 14105	0.2 max		0.06
Triglyceride	% mass	EN 14105	0.2 max		<0.001
Free Glycerol	% mass	EN 14105	0.02 max	0.02 max	0.005
Total Glycerol	% mass	EN 14105	0.25 max	0.24 max	0.183
Iodine Value		EN 14111	120 max		127
Linolenic Acid Methyl Ester	% mass	EN 14103	12 max		5.9
Polyunsaturated Methyl Esters	% mass		1 max		
Phosphorous	mg/kg	ASTM D4951	10 max	10 max	<0.1
Alkalinity	mg/kg				
Group I Metals [Na, K]	mg/kg	EN 14109	5 max		<0.1
Group II Metals [Ca, Mg]	mg/kg	EN 14538	5 max		<0.1

# Glycerin Quality Results & discussion

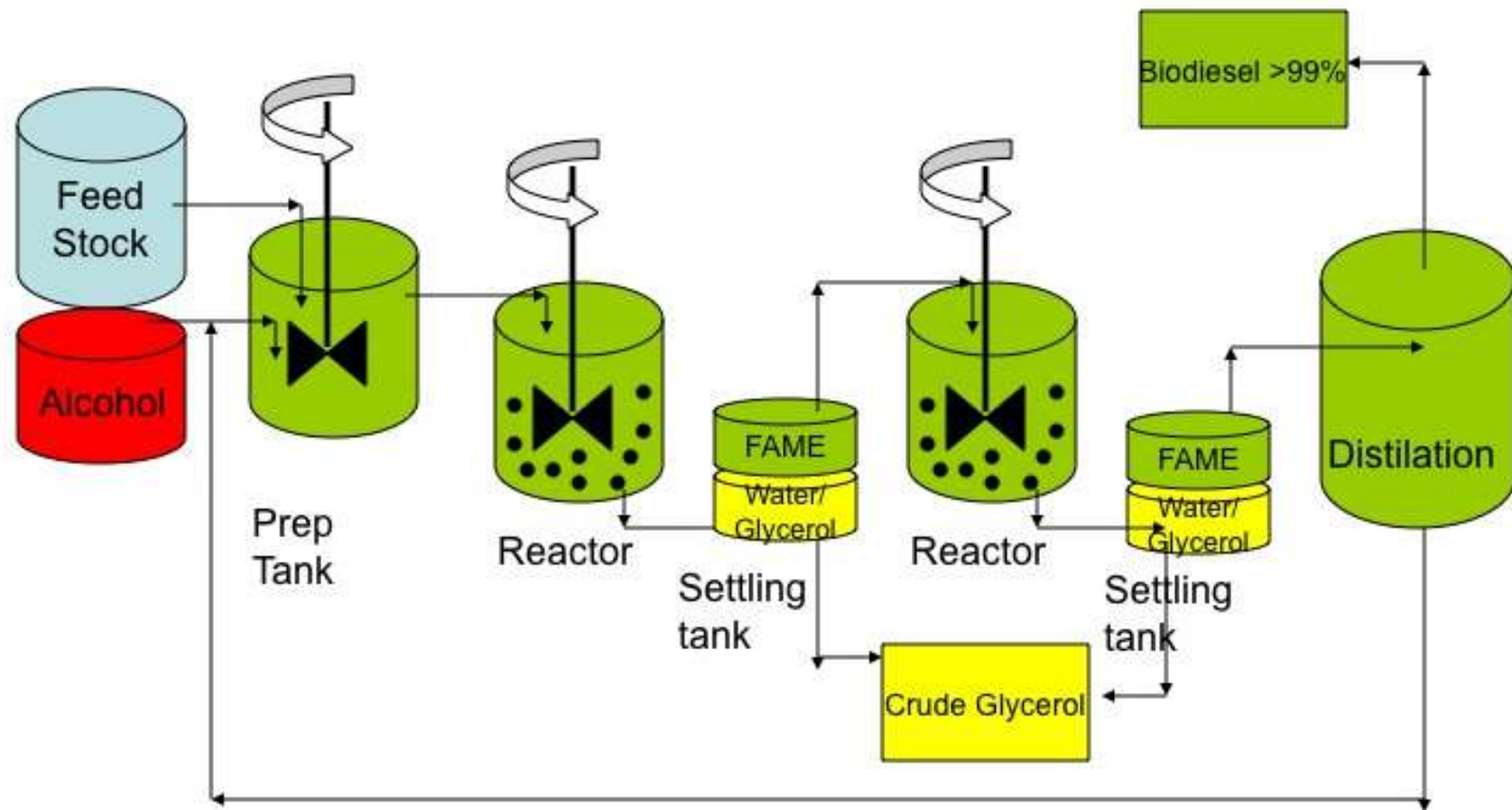
Quality analysis of enzymatically formed glycerin vs chemically formed glycerin

Chemically	Enzymatically	Property
Very brown	Transparent	Appearance
44% - 54%	78%	Glycerin content
8.9 - 10.5	6.0 - 6.2	pH
27 - 53%	3%	Methanol content
1 - 4%	7%	Water content
7%	0.1%	Sulphated ash



The results of the analysis confirmed that the enzymatically formed glycerin is of much better quality than the glycerin formed via conventional chemical process

# Proposed State of Art Enzymatic Process Overview (1)

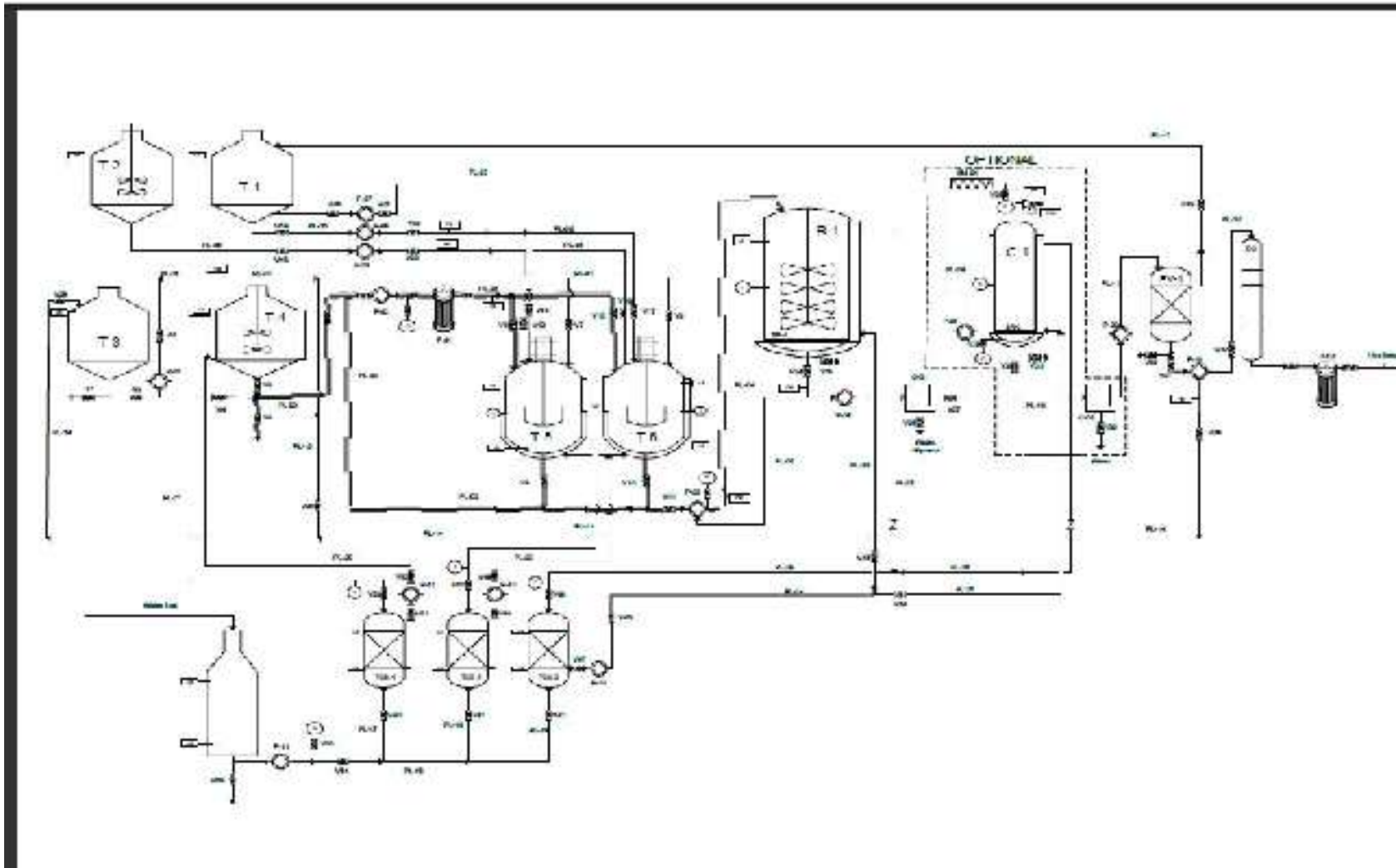




# Proposed State of Arts Enzymatic Process Overview

(2)

P&I diagram example





# Enzymatic Biodiesel Plant Running Examples (15 ton/day)



# Enzymatic Biodiesel Plant underconstruction Example (35 ton/day)



# Proposed EnZymatic Process Summary

We believe the proposed EnZymatic process is:

- the most advanced state of art & cutting edge technology and also being field proven biodiesel process.
- used for the most wide and multi-feedstocks(0–100%FFA,virtually all feedstocks available), as well as waste vegetable oil(Ca'nt find elsewhere !).
- the highest purity biodiesel fuel as well as glycerin are obtained.
- the easiest plant operation and simplest process flow .
- the most cost competitive and economical process and can keep your competitiveness & investment for long time future.
- Our Japanese manufacturing excellence&quality control(QC) standard&practices and extensive knowledge&expertise in biodiesel.

End of Presentation

Thank you  
by Akira Hirai